

2-(1*H*-Benzotriazol-1-yl)-1-(2,4-dichlorophenyl)-
2-(1*H*-1,2,4-triazol-1-yl)ethanoneJun Wan,^a Chun-Li Li,^a Xue-Mei Li,^a Shu-Sheng Zhang^{a*} and Ping-Kai Ouyang^b^aCollege of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, 266042 Qingdao, Shandong, People's Republic of China, and ^bCollege of Life Science and Pharmaceutical Engineering, Nanjing University of Technology, 210093 Nanjing, Jiangsu, People's Republic of China

Correspondence e-mail: shushzhang@126.com

In the title compound, C₁₆H₁₀Cl₂N₆O, molecules are linked into ribbons along the *b* axis by C—H···N intermolecular hydrogen bonds. The packing is further stabilized by a π – π interaction.

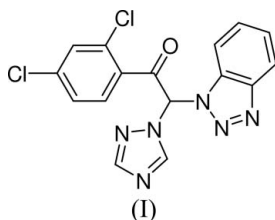
Received 6 September 2005

Accepted 9 September 2005

Online 14 September 2005

Comment

Triazole derivatives exhibit growth-inhibiting activities against some microorganisms (Xu *et al.*, 2002). In order to search for new triazole compounds with higher bioactivity, the title compound, (I), which contains both triazole and benzotriazole, was synthesized.



Key indicators

Single-crystal X-ray study

T = 293 KMean σ (C—C) = 0.004 Å*R* factor = 0.053*wR* factor = 0.130

Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The bond lengths and angles in (I) are within normal ranges (Allen *et al.*, 1987) and comparable to those reported in the related compound 2-(1*H*-1,2,3-benzotriazol-1-yl)-1-(4-methylphenyl)-2-(1*H*-1,2,4-triazol-1-yl)ethanone (Wan *et al.*, 2005). The benzotriazole group is essentially planar, with a dihedral angle of 1.9 (1)° between the benzene and triazole rings. The mean plane of the benzotriazole group makes dihedral angles of 74.3 (1) and 33.9 (1)° with the other triazole (N4/N5/C16/N6/C15) ring and the benzene (C1–C6) ring, respectively. The dihedral angle between the planes of the latter two aromatic rings is 69.9 (1)°. In the crystal structure, molecules are linked into ribbons by intermolecular C—H···N hydrogen bonds (Fig. 2 and Table 2). The packing is further stabilized by a π – π interaction between the benzene (C1–C6) rings, the distance between the centroids [*Cg*···*Cg*($\frac{1}{2} - x, \frac{3}{2} - y, -z$)] being 3.883 (2) Å.

Experimental

The title compound was prepared according to the method of Wan *et al.* (2005).

Crystal data

C₁₆H₁₀Cl₂N₆O*M_r* = 373.20Monoclinic, *C*2/*c**a* = 21.950 (3) Å*b* = 9.9632 (15) Å*c* = 15.538 (2) Å β = 106.028 (2)°*V* = 3266.0 (8) Å³*Z* = 8*D_x* = 1.518 Mg m^{−3}Mo *K*α radiation

Cell parameters from 2583

reflections

 θ = 2.3–24.0° μ = 0.42 mm^{−1}*T* = 293 (2) K

Column, colorless

0.40 × 0.23 × 0.12 mm

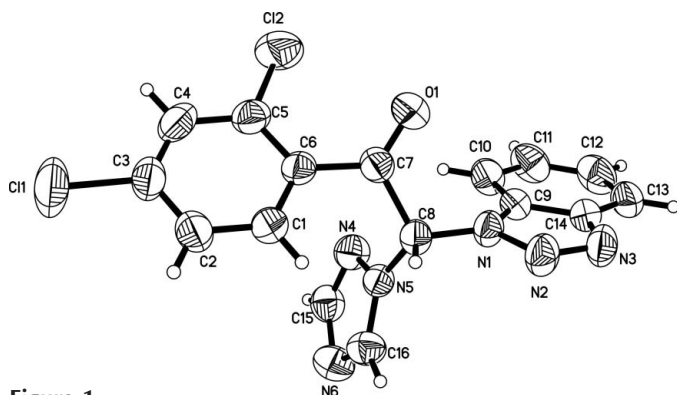


Figure 1
The structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

Data collection

Siemens SMART 1000 CCD area-detector diffractometer	3199 independent reflections
ω scans	2500 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.023$
$T_{\text{min}} = 0.851$, $T_{\text{max}} = 0.952$	$\theta_{\text{max}} = 26.0^\circ$
8949 measured reflections	$h = -27 \rightarrow 22$
	$k = -11 \rightarrow 12$
	$l = -18 \rightarrow 19$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0519P)^2 + 3.9871P]$
$R[F^2 > 2\sigma(F^2)] = 0.053$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.130$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.42 \text{ e } \text{\AA}^{-3}$
3199 reflections	$\Delta\rho_{\text{min}} = -0.36 \text{ e } \text{\AA}^{-3}$
226 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (\AA , $^\circ$).

Cl1—C3	1.731 (3)	N1—C8	1.445 (3)
Cl2—C5	1.723 (3)	N5—C8	1.447 (3)
O1—C7	1.201 (3)	C7—C8	1.549 (3)
N1—C8—N5	111.83 (19)	N5—C8—C7	110.26 (18)
N1—C8—C7	112.21 (19)		

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C1—H1A \cdots N3 ⁱ	0.93	2.54	3.465 (3)	172
C8—H8A \cdots N3 ⁱ	0.98	2.54	3.455 (3)	156

Symmetry code: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, -z - \frac{1}{2}$.

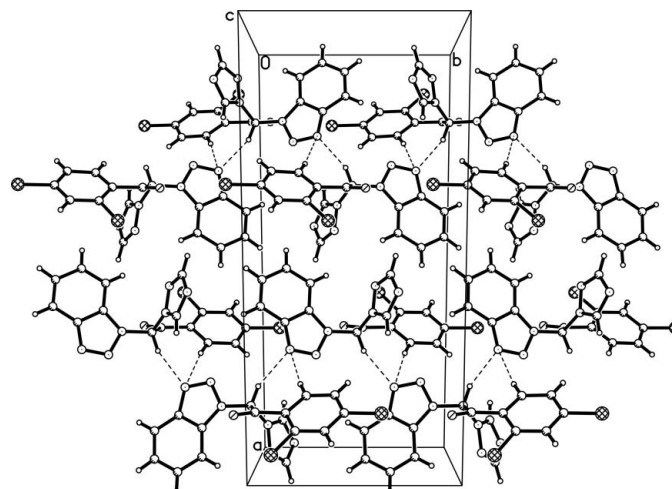


Figure 2

A view down the c axis. Hydrogen bonds are indicated by dashed lines.

All H atoms were located in difference Fourier maps and constrained to ride on their parent atoms, with C—H distances in the range 0.93–0.98 \AA and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Siemens, 1996); software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

This project was supported by the Program for New Century Excellent Talents in Universities (No. NCET-04-0649) and the Project of Educational Administration of Shandong Province (No. J04B12).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Siemens (1996). SMART, SAINT and SHELXTL. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Wan, J., Li, C.-L., Li, X.-M., Ouyang, P.-K. & Zhang, S.-S. (2005). *Chem. Res. Chin. Univ.* Submitted.
- Xu, L. Z., Jiao, K., Zhang, S. S. & Kuang, S. P. (2002). *Bull. Korean Chem. Soc.* **23**, 1699–1701.